



ORIGINAL ARTICLE

# Characterization and antibacterial activity of nanocrystalline Mn doped Fe<sub>2</sub>O<sub>3</sub> thin films grown by successive ionic layer adsorption and reaction method



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**Abstract** Successive ionic layer adsorption and reaction (SILAR) method have been successfully employed to grow nanocrystalline Mn doped  $\alpha$ -Fe<sub>2</sub>O<sub>3</sub> thin films onto glass substrates. The structural analysis revealed that, the films are nanocrystalline in nature with rhombohedral structure. The optical studies showed that  $\alpha$ -Fe<sub>2</sub>O<sub>3</sub> thin film exhibits 3.02 eV band gap energy and it decreases to 2.95 eV as the Mn doping percentage in it was increased from 0 to 8 wt.%. The SILAR grown  $\alpha$ -Fe<sub>2</sub>O<sub>3</sub> film exhibits antibacterial character against *Staphylococcus aureus* bacteria and it increases remarkably with Mn doping.

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## 1. Introduction

Since the past decades,  $\alpha$ -Fe<sub>2</sub>O<sub>3</sub> has gained extensive scientific importance in materials science because of its important role in various applications namely gas sensor, supercapacitor, dye sensitized solar cell, photocatalyst, lithium ion battery and in microbial fuel cells (Lee et al., 2001; Fan et al., 2011;

Mulenko et al., 2012; Rahman and Joo, 2013; Cavas et al., 2013; Akhavan, 2010; Hsien et al., 2013; Kitaura et al., 2008; Kulal et al., 2011; Ji et al., 2011). In recent years, many researchers have studied the role of doping in  $\alpha$ -Fe<sub>2</sub>O<sub>3</sub> to improve its applicability for electrochemical sensors, solid oxide fuel cell and photo splitting of water etc. (Suresh et al., 2012; Geng et al., 2012; Shwarsstein et al., 2008). In addition, Shinde et al. (2011) have reported structural, morphological, luminescent and electronic properties of spray deposited Al doped  $\alpha$ -Fe<sub>2</sub>O<sub>3</sub> thin films. Khan and Zhou (1993) have reported physical properties of iodine doped  $\alpha$ -Fe<sub>2</sub>O<sub>3</sub> thin films grown by spray pyrolysis. Sanchez et al. (1988) have

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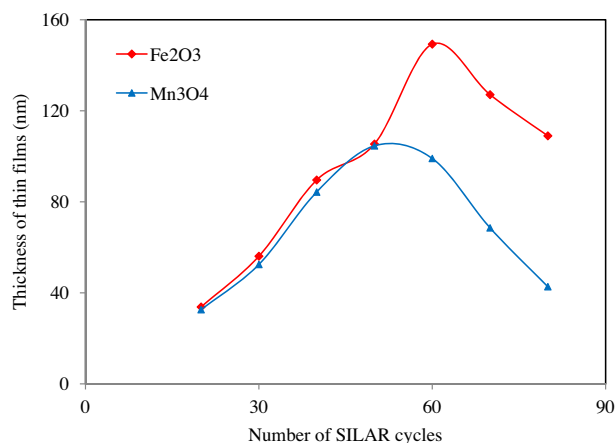
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reported photoelectrochemical properties of niobium doped  $\alpha$ -Fe<sub>2</sub>O<sub>3</sub> thin films grown by chemical vapor deposition method. Sensing properties of LPD synthesized Pd and Ca doped  $\alpha$ -Fe<sub>2</sub>O<sub>3</sub> thin films were reported by Neri et al. (2007). As per literature to grow doped and undoped  $\alpha$ -Fe<sub>2</sub>O<sub>3</sub> thin films, various chemical deposition methods have been utilized viz. spin coating deposition (Souza et al., 2009), liquid-phase method (Neri et al., 2001), electrodeposition (Kumar et al., 2011), spray-pyrolysis (Kumari et al., 2010) and DC sputtering (Kiran et al., 2006). The growth of nanostructured thin films by a simple and economic deposition technique has been playing an important role in the field of nanoscience and nanotechnology in order to reduce the device fabrication cost. In this paper, we have outlined the simple and economic successive ionic layer adsorption and reaction method to grow nanocrystalline Mn doped  $\alpha$ -Fe<sub>2</sub>O<sub>3</sub> thin films. This is an excellent method to grow a nanocrystalline thin film by immersing the glass substrate into separately placed cationic and anionic precursors alternately. In between each immersion, the substrate is rinsed in deionized water to remove the loosely bound species to get an adherent thin film. As cationic and anionic precursors are separately placed, it is very easy to control the growth process by adjusting the number of deposition cycles. The structural, morphological and optical properties of nanocrystalline Mn doped  $\alpha$ -Fe<sub>2</sub>O<sub>3</sub> thin films grown by the SILAR method are discussed in the paper. Also, the results of antibacterial activity of the Mn doped  $\alpha$ -Fe<sub>2</sub>O<sub>3</sub> films against *Staphylococcus aureus* bacteria are reported.

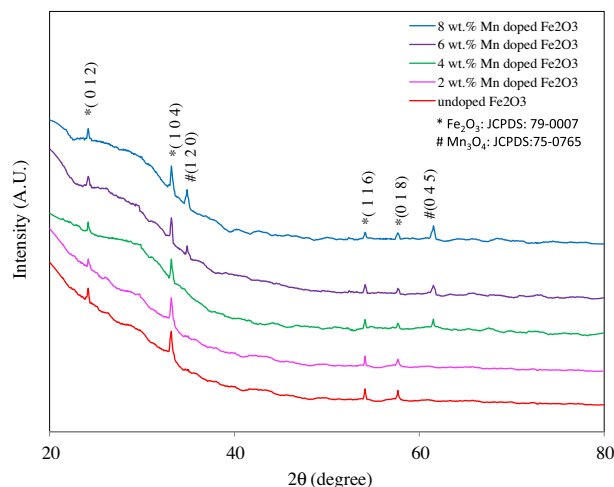
## 2. Experimental

### 2.1. Preparation of Mn doped $\alpha$ -Fe<sub>2</sub>O<sub>3</sub> thin films

To grow Mn doped  $\alpha$ -Fe<sub>2</sub>O<sub>3</sub> thin films 0.3 M MnCl<sub>2</sub> of pH 1 and 0.06 M FeCl<sub>2</sub> of pH < 1 were used as cationic precursors along with 0.005 M NaOH (pH  $\approx$  12) solution as an anionic precursor. The beakers containing precursor solutions were placed alternately in such a way that, every cationic and anionic precursor was followed by a beaker containing deionized water. Several deposition trials were performed to optimize the various deposition parameters to grow Mn<sub>3</sub>O<sub>4</sub> and  $\alpha$ -Fe<sub>2</sub>O<sub>3</sub> films separately. For  $\alpha$ -Fe<sub>2</sub>O<sub>3</sub> film, a well cleaned glass substrate was immersed in a cationic precursor for 20 s where Fe<sup>3+</sup> ions were adsorbed on the substrate surface. The substrate was then rinsed in deionized water for 20 s to remove loosely bound Fe<sup>3+</sup> ions. Finally it was immersed in NaOH solution for 20 s, where OH<sup>-</sup> ions react with Fe<sup>3+</sup> ions to form  $\alpha$ -Fe<sub>2</sub>O<sub>3</sub> species. This was again followed by rinsing in deionized water for 20 s to remove loose material from the substrate surface. This completes one SILAR deposition cycle for  $\alpha$ -Fe<sub>2</sub>O<sub>3</sub>. The immersion and rinsing time periods were experimentally optimized to get uniform and adherent thin films. The optimized deposition parameters for Mn<sub>3</sub>O<sub>4</sub> films were already explained in our earlier report (Ubale et al., 2012). To achieve proper doping, several deposition trials were performed by varying the concentration of Fe precursor to match the growth rate of  $\alpha$ -Fe<sub>2</sub>O<sub>3</sub> film formation with Mn<sub>3</sub>O<sub>4</sub>. It was observed that, the growth rate of  $\alpha$ -Fe<sub>2</sub>O<sub>3</sub> and Mn<sub>3</sub>O<sub>4</sub> is approximately the same for 0.06 and 0.3 M concentrations of FeCl<sub>3</sub> and MnCl<sub>2</sub> respectively. Fig. 1 shows the variation of Mn<sub>3</sub>O<sub>4</sub> and  $\alpha$ -Fe<sub>2</sub>O<sub>3</sub> film thickness with a number of deposition cycles. It was observed



**Figure 1** Plot of  $\alpha$ -Fe<sub>2</sub>O<sub>3</sub> and Mn<sub>3</sub>O<sub>4</sub> film thickness with number of SILAR cycles [Cationic Source: 0.06 M FeCl<sub>3</sub> (for Fe<sub>2</sub>O<sub>3</sub>) and 0.3 M MnCl<sub>2</sub> (for Mn<sub>3</sub>O<sub>4</sub>)].



**Figure 2** GIXRD patterns of Mn doped  $\alpha$ -Fe<sub>2</sub>O<sub>3</sub> thin films.

**Table 1** Film thickness and crystallite size of  $\alpha$ -Fe<sub>2</sub>O<sub>3</sub> with Mn doping percentage.

| wt.% of Mn in $\alpha$ -Fe <sub>2</sub> O <sub>3</sub> | Film thickness (nm) | Average crystallite size from |            |
|--|---------------------|-------------------------------|------------|
|  |                     | GIXRD (nm)                    | FESEM (nm) |
| 0  | 105                 | 21                            | 28         |
| 2  | 108                 | 21                            | 26         |
| 4  | 107                 | 20                            | 23         |
| 6  | 105                 | 19                            | 21         |
| 8  | 106                 | 17                            | 20         |

that up to 50 SILAR cycles the growth rate was approximately the same. Above 50 SILAR cycles, thickness of Mn<sub>3</sub>O<sub>4</sub> film decreases as it peels off; however, the thickness of  $\alpha$ -Fe<sub>2</sub>O<sub>3</sub> film increases up to 60 SILAR cycles. Hence for the present work, to grow Mn doped  $\alpha$ -Fe<sub>2</sub>O<sub>3</sub> thin films 50 SILAR cycles were considered. To achieve 0, 2, 4, 6 and 8 wt.% doping of Mn in  $\alpha$ -Fe<sub>2</sub>O<sub>3</sub>, the number of SILAR cycles for (Fe<sub>2</sub>O<sub>3</sub>:Mn<sub>3</sub>O<sub>4</sub>) composition were taken as (50:0), (49:1), (48:2), (47:3) and (46:4) respectively. These films were further annealed at 500<sup>o</sup> C for

3 h to get a pure metal oxide phase of the deposited material and used for further characterization.

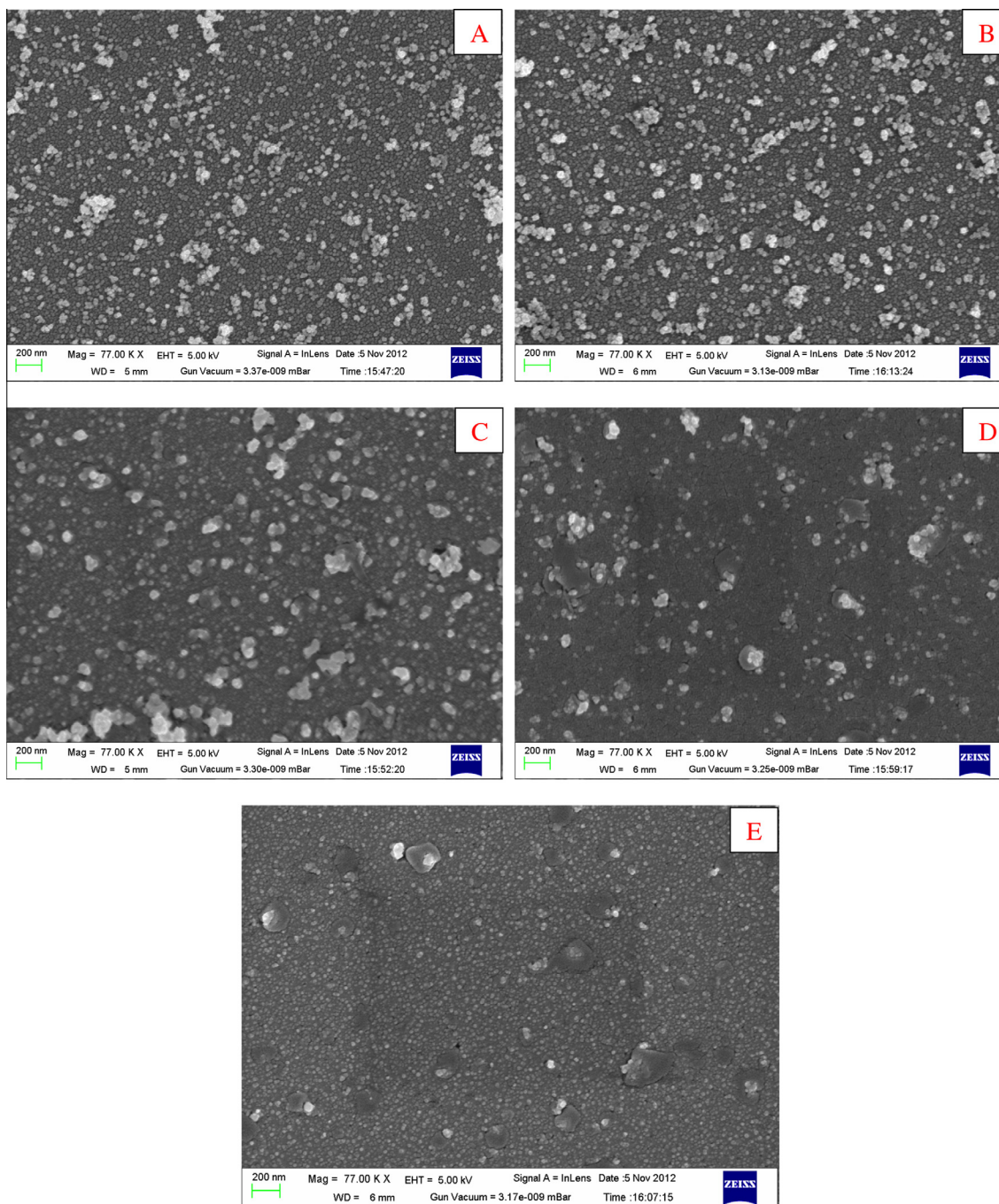
## 2.2. Characterization techniques

In the present work, thickness of the film was measured by gravimetric weight difference method by using the relation,

$$t = \frac{m}{\rho \times A} \quad (1)$$

where ' $m$ ' is the mass of the deposited material measured by using a sensitive microbalance; ' $A$ ' is the area of the deposited

film in  $\text{cm}^2$  and ' $\rho$ ' is the density of the deposited material. It was observed that, the thickness of Mn doped  $\alpha\text{-Fe}_2\text{O}_3$  thin films deposited by repeating 50 SILAR cycles is of the order of 105 nm. The crystal structure of the deposited film was identified by grazing incidence X-ray diffraction with Xpert PRO PANalytical diffractometer. The surface morphological investigations were carried out by using field emission scanning electron microscope (Model: SUPRA 40) and atomic force microscope (Model: Nanonics Multiview 2000<sup>TM</sup>, Israel). The optical absorption studies were carried out in the wavelength range 350–750 nm using ELICO<sup>®</sup> Double Beam SL 210 UV–Vis spectrophotometer.



**Figure 3** FESEM images of Mn doped  $\alpha\text{-Fe}_2\text{O}_3$  thin films; Mn doping percentage: (A) 0 wt.%, (B) 2 wt.%, (C) 4 wt.%, (D) 6 wt.% and (E) 8 wt.%.

### 2.3. Antibacterial test

A spread plate technique was employed to investigate antibacterial behavior of Mn doped  $\alpha$ -Fe<sub>2</sub>O<sub>3</sub> thin films against *S. aureus*. For it, the culture of *S. aureus* bacteria was prepared in nutrient broth. The loopful culture of *S. aureus* organisms was further inoculated into 20 mL sterilized nutrient broth and incubated at 310 K for 24 h. Then, 20  $\mu$ L cultures of *S. aureus* was inoculated on undoped and Mn doped iron oxide glass substrates of area 1 cm<sup>2</sup> with the help of an inoculating needle. These glass slides were then placed in previously sterilized petri dishes and incubated at 310 K for 24 h. After incubation these slides were transferred to 3 mL of buffer peptone solution in a test tube and ultrasonicated to detach the bacteria thoroughly from the substrate. From this, 20  $\mu$ L washed buffer peptone solution was then inoculated on nutrient agar plates by a spread plate technique and incubated at 310 K for 24 h to obtain viable bacteria. After successful incubation the viable bacterial colonies were counted and antibacterial efficiency was calculated using the relation (Zhang et al., 2008),

$$r = \frac{(N_0 - N)}{N_0} \times 100\% \quad (2)$$

where, 'r' is the antibacterial efficiency, 'N<sub>0</sub>' is the number of viable bacterial colonies observed in the petri dish from undeposited film and 'N' is the number of viable bacterial colonies observed in the petri dish from Mn doped Fe<sub>2</sub>O<sub>3</sub> thin film.

## 3. Results and discussion

### 3.1. Structural studies

Fig. 2 shows typical GIXRD patterns of Mn doped iron oxide thin films. The pattern revealed that SILAR grown Fe<sub>2</sub>O<sub>3</sub> films are nanocrystalline in nature with rhombohedral structure. The observed GIXRD data are in good agreement with the standard data [JCPDS: 79-0007 and JCPDS: 75-0765]. The (012), (104), (116) and (018) orientations observed in the patterns are due to  $\alpha$ -Fe<sub>2</sub>O<sub>3</sub>, which is in good agreement with the results reported by several workers (Akhavan, 2010; An et al., 2009; Cha et al., 2009). For higher wt.% of Mn doping (0.45) and (120) peaks due to Mn<sub>3</sub>O<sub>4</sub> are observed. It is also observed that the intensity of diffraction peaks due to Mn<sub>3</sub>O<sub>4</sub> increases with doping percentage. The average crystallite size of the deposited material was determined by using Debye–Scherrer formula (Ubale et al., 2013),

$$D = \frac{0.9\lambda}{\beta \cos\theta} \quad (3)$$

where ' $\lambda$ ' is the wavelength used (0.154 nm); ' $\beta$ ' is the angular line width at half maximum intensity in radians and ' $\theta$ ' is the Bragg's angle. It is found that the crystallite size of the  $\alpha$ -Fe<sub>2</sub>O<sub>3</sub> film is of the order of 21 nm, and it decreases to 17 nm as doping percentage of Mn was increased from 0 to 8 wt.% (Table 1).

### 3.2. Surface morphology

The surface morphology of the iron oxide thin films deposited onto glass substrates by the SILAR method was examined by using field emission scanning electron micrographs (Fig. 3).

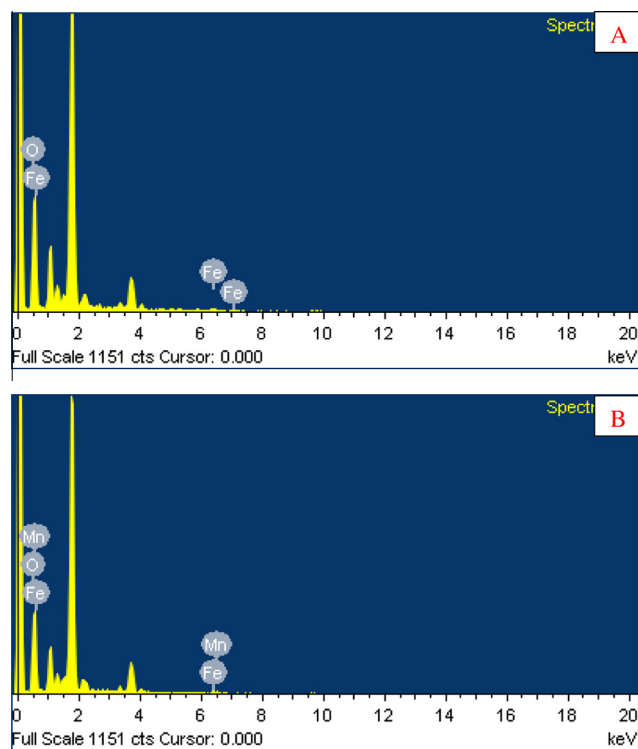
The FESEM analysis shows compact and homogeneous distribution of nano grains of varying sizes from 28 to 20 nm depending upon doping percentage. The agglomeration of nano grains is observed at several places on the film surface. It is seen that the number of agglomerated grains reduces with Mn doping. The grain sizes estimated from FESEM images are in good agreement with GIXRD results (Table 1).

The elemental analysis of the undoped and 6 wt.% Mn doped  $\alpha$ -Fe<sub>2</sub>O<sub>3</sub> thin films were carried out using energy dispersive X-ray (EDX) analysis (Fig. 4). The elemental analysis was carried out only for Fe, O and Mn elements. The additional peaks observed in the spectra are due to the composition of the glass substrate. The small minor peak of Mn observed in the spectra confirms its doping in  $\alpha$ -Fe<sub>2</sub>O<sub>3</sub>.

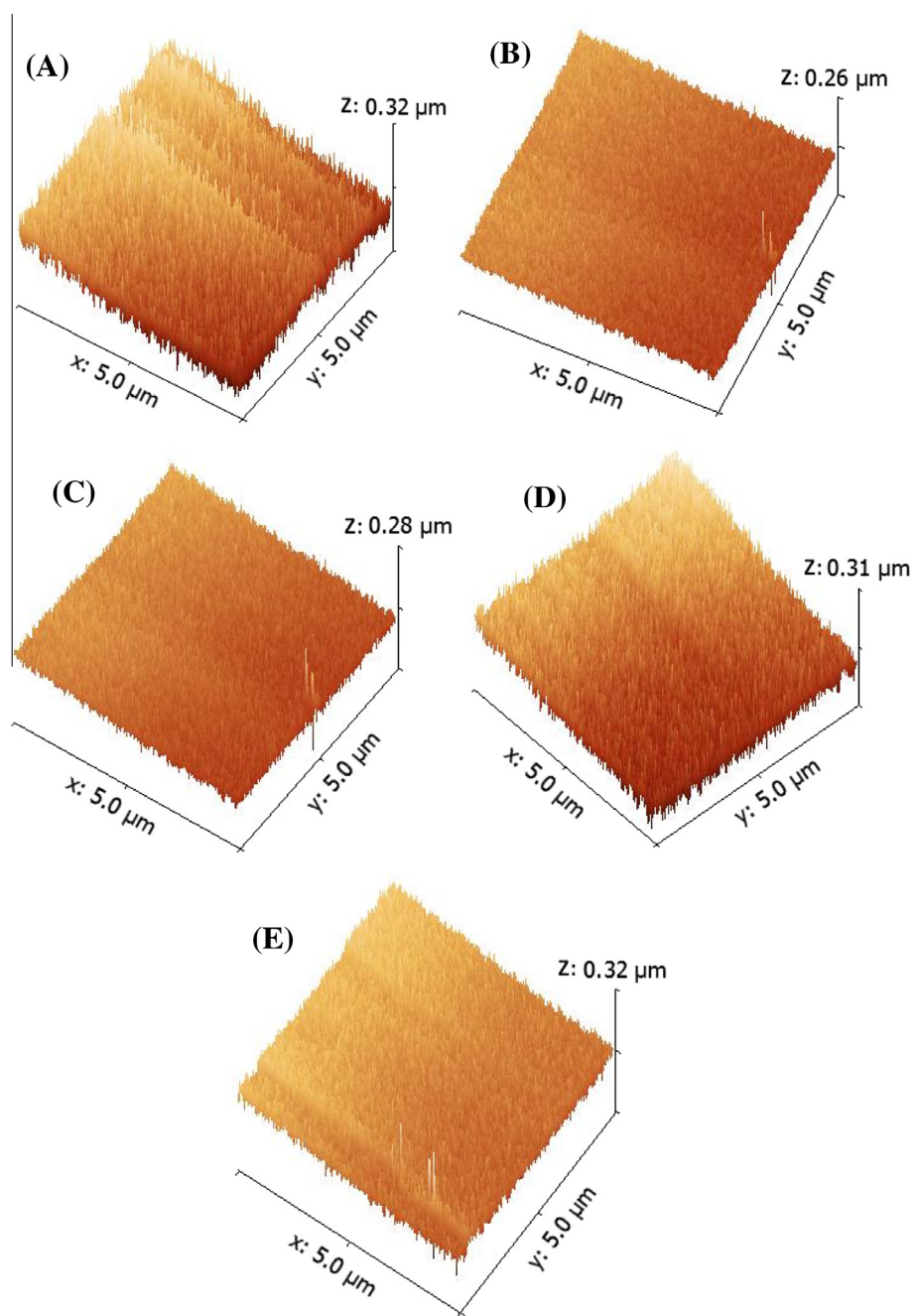
Fig. 5 shows 3D AFM images of Mn doped  $\alpha$ -Fe<sub>2</sub>O<sub>3</sub> thin films. The images were recorded in tapping mode by using an optical fiber tip, coated with gold and chromium (Au, Cr) metal with  $\phi = 20$  nm at response frequency 52.38 kHz. The AFM analysis showed that the granular structure of  $\alpha$ -Fe<sub>2</sub>O<sub>3</sub> is uniformly distributed over the entire substrate surface. The morphological parameters such as rms roughness (Rq), average roughness (Ra), average height, maximum height and grain orientation depending on doping percentage of Mn are listed in Table 2. The rms roughness of  $\alpha$ -Fe<sub>2</sub>O<sub>3</sub> film decreases as doping concentration of Mn rises from 0 to 8 wt.%. The rms roughness and average roughness becomes approximately equal above 6 wt.% doping of Mn which may be because of uniform mixing of Mn<sub>3</sub>O<sub>4</sub> in  $\alpha$ -Fe<sub>2</sub>O<sub>3</sub>.

### 3.3. Optical studies

The optical absorption measurements for  $\alpha$ -Fe<sub>2</sub>O<sub>3</sub> thin films deposited onto glass substrates were carried out in the wavelength



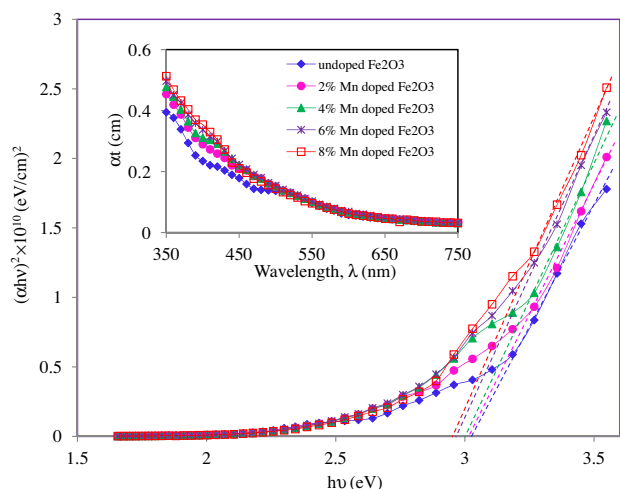
**Figure 4** EDAX spectrum (A) as deposited Fe<sub>2</sub>O<sub>3</sub> film and (B) 6 wt.% Mn doped  $\alpha$ -Fe<sub>2</sub>O<sub>3</sub> film.



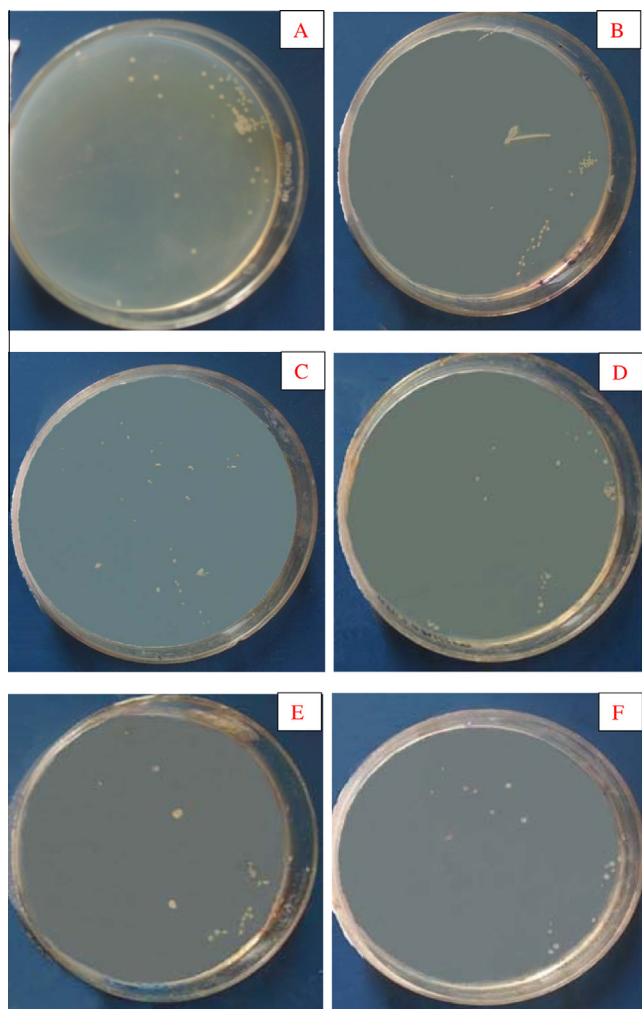
**Figure 5** 3D AFM images of Mn doped  $\alpha$ -Fe<sub>2</sub>O<sub>3</sub> thin films; Mn doping percentage: (A) 0 wt.%, (B) 2 wt.%, (C) 4 wt.%, (D) 6 wt.% and (E) 8 wt.%.

**Table 2** Morphological parameters of Mn doped  $\alpha$ -Fe<sub>2</sub>O<sub>3</sub> thin films.

| wt.% of Mn in Fe <sub>2</sub> O <sub>3</sub> | RMS roughness Rq (nm) | Average surface roughness Ra(nm) | Maximum height (nm) | Average height (nm) | Grain orientation (pi) |
|--|-----------------------|----------------------------------|---------------------|---------------------|------------------------|
| 0  | 86.01                 | 71.83                            | 323.2               | 121.72              | 0.07                   |
| 2  | 77.53                 | 62.14                            | 267.5               | 118.01              | 0.16                   |
| 4  | 71.05                 | 59.82                            | 280.6               | 117.34              | 0.08                   |
| 6  | 68.72                 | 57.45                            | 317.4               | 114.3               | 0.08                   |
| 8  | 66.15                 | 54.03                            | 326.6               | 112.78              | 0.12                   |



**Figure 6** The plots of  $(\alpha hv)^2$  versus  $hv$  (Inset plots of optical absorption versus wavelength) of Mn doped  $\alpha$ -Fe<sub>2</sub>O<sub>3</sub> thin films.



**Figure 7** Antibacterial test results of *S. aureus* after 24 h: incubated on (A) undeposited glass substrate and on Mn doped  $\alpha$ -Fe<sub>2</sub>O<sub>3</sub> thin films with Mn doping (B) 0 wt.%, (C) 2 wt.%, (D) 4 wt.%, (E) 6 wt.% and (F) 8 wt.%.

range 350–750 nm at room temperature shown in the inset of Fig. 6. An increase in the optical absorption of  $\alpha$ -Fe<sub>2</sub>O<sub>3</sub> films was observed with increase in doping percentage of Mn. The optical band gap energies ( $E_g$ ) of undoped and Mn doped  $\alpha$ -Fe<sub>2</sub>O<sub>3</sub> films were calculated by using the equation (Mahdi et al., 2012),

$$\alpha hv = A(hv - E_g)^n \quad (4)$$

where, ' $\alpha$ ' is absorption coefficient, ' $E_g$ ' is band gap energy, ' $A$ ' is a constant and ' $n$ ' is equal to 1/2 for direct and 2 for indirect transition. Fig. 6 illustrates the plots of  $(\alpha hv)^2$  versus  $hv$  for Mn doped  $\alpha$ -Fe<sub>2</sub>O<sub>3</sub> thin films. The band gap energy ' $E_g$ ' of the  $\alpha$ -Fe<sub>2</sub>O<sub>3</sub> film was estimated by extrapolating the linear portion of the plot to the energy axis and is found to be of the order of 3.02 eV. This estimated optical band gap energy of the  $\alpha$ -Fe<sub>2</sub>O<sub>3</sub> thin film is in good agreement with the results reported by Glasscock et al. (2008) and Bhar et al. (2010). However, it is observed that, the optical band gap energy of the film decreases from 3.02 to 2.95 eV as Mn doping in  $\alpha$ -Fe<sub>2</sub>O<sub>3</sub> rises from 0 to 8 wt.%.

### 3.4. Antibacterial activity

The antibacterial character of the Mn doped  $\alpha$ -Fe<sub>2</sub>O<sub>3</sub> thin films were investigated against *S. aureus* bacteria. The test results of *S. aureus* bacteria incubated at 24 h on undeposited glass substrate and Mn doped  $\alpha$ -Fe<sub>2</sub>O<sub>3</sub> thin film surfaces are shown in Fig. 7. It was observed that, the antibacterial efficiency for undoped  $\alpha$ -Fe<sub>2</sub>O<sub>3</sub> film is 16.66% and it increases to 58.33% as doping of Mn increases from 0 to 8 wt.%. This increased antibacterial efficiency may be due to the improved nanocrystalline nature of  $\alpha$ -Fe<sub>2</sub>O<sub>3</sub> films. But, the interactions of nanoparticles with bacteria are dependent on the size, shape and morphology of the deposited material (Yu et al., 2011; Sikong et al., 2010). When these nanoparticles come in contact with a bacterial cell, an active oxygen is formed due to the chemisorption process, as a result more number of iron, manganese and hydroxide ions or hydrogen peroxide were released from the surface, which can react with the peptide linkages in the cell wall of bacteria and disrupt them. This antibacterial mechanism may be involved in the antibacterial study to kill the *S. aureus* microorganisms (Zhang et al., 2008; Touati, 2000; Keenan and Sedlak, 2008; Ubale and Belkhedkar, 2015; Belkhedkar and Ubale, 2014; Aninwene et al., 2013). It was also observed from the AFM images that the rms roughness of  $\alpha$ -Fe<sub>2</sub>O<sub>3</sub> film decreases with the increase of Mn doping, which plays a very important role in the antibacterial activity against the *S. aureus* bacteria. Due to decreased rms roughness of the film, more nanoparticles of deposited material interacted with the bacteria and damaged the cell wall which further blocks the bacterial production. As the rms roughness of the  $\alpha$ -Fe<sub>2</sub>O<sub>3</sub> thin films is almost closure above 4 wt.% doping of Mn, its antibacterial activity against *S. aureus* bacteria becomes nearly constant. Similarly, the role of rms roughness on the antibacterial activity of nano-barium sulfate incorporated pellethane bio-film against *S. aureus* and *Pseudomonas aeruginosa* bacteria was reported by Aninwene et al. (2013).

## 4. Conclusions

In the present work, Mn doped  $\alpha$ -Fe<sub>2</sub>O<sub>3</sub> thin films were successfully synthesized by successive ionic layer adsorption and reaction method onto glass substrates. The GIXRD,

FESEM, EDX and AFM characterizations confirm that as grown Mn doped  $\alpha$ -Fe<sub>2</sub>O<sub>3</sub> thin films are nanocrystalline in nature. The optical band gap energy of  $\alpha$ -Fe<sub>2</sub>O<sub>3</sub> thin film increases with the doping percentage of Mn. The optical band gap energy and film morphology can be engineered by adjusting the doping of Mn in  $\alpha$ -Fe<sub>2</sub>O<sub>3</sub>, so as to make them suitable for optoelectronic devices. Also, the antibacterial character of  $\alpha$ -Fe<sub>2</sub>O<sub>3</sub> against *S. aureus* bacteria enhances remarkably with Mn doping.

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